

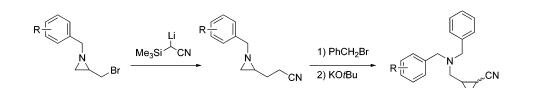
Synthesis of 1-Arylmethyl-2-(2-cyanoethyl)aziridines and Their Rearrangement into Novel 2-(Aminomethyl)cyclopropanecarbonitriles

Matthias D'hooghe, Karel Vervisch, and Norbert De Kimpe*

Department of Organic Chemistry, Faculty of Bioscience Engineering, Ghent University, Coupure Links 653, B-9000 Ghent, Belgium

norbert.dekimpe@UGent.be

Received June 19, 2007



1-Arylmethyl-2-(bromomethyl)aziridines were transformed into novel 2-(2-cyanoethyl)aziridines upon treatment with α -lithiated trimethylsilylacetonitrile in THF in an efficient and straightforward approach. The latter aziridines underwent ring opening by reaction with benzyl bromide in acetonitrile, affording 5-amino-4-bromopentanenitriles through a regiospecific ring opening of intermediate aziridinium salts by bromide. Further elaboration of these γ -bromonitriles resulted in the synthesis of novel 2-[*N*,*N*-bis-(arylmethyl)aminomethyl]cyclopropanecarbonitriles in high yields by means of a 1,3-cyclization protocol upon treatment with KOtBu in THF.

Introduction

In recent years, many efforts have been devoted to the development of new methods for the biocatalytic conversion of aminonitriles into the corresponding amino acids.¹ Consequently, the search for novel types of functionalized aminonitrile derivatives has gained much interest and has become an important challenge in organic synthesis.² Furthermore, conformationally constrained amino acids have attracted special attention because of their potential usefulness in medicinal chemistry,³ although synthetic approaches toward these compounds or their precursors are rather scarce. In continuation of our interest in the preparation of 2-aminocyclopropanecarboxy-lates (β -ACC's)⁴ and 2-aminocyclopropanecarbonitriles,⁵ the

synthesis of 2-(aminomethyl)cyclopropanecarbonitriles was envisaged as constrained analogues of the neurotransmitter γ -aminobutyric acid (GABA). 2-(Aminomethyl)cyclopropanecarboxylates **1** are considered to be valuable compounds in medicinal chemistry with a broad applicability, ranging from GABA_C (ant)agonists⁶ to a potential use in the treatment of diabetes II,⁷ cancer,⁸ and hepatitis C.⁹ Also in peptide chemistry, interesting applications of analogous compounds have been reported, such as the recent discovery of a parallel sheet structure

 ^{(1) (}a) Fitz, M.; Lundell, K.; Lindroos, M.; Fülöp, F.; Kanerva, L. T. Tetrahedron: Asymmetry 2005, 16, 3690. (b) Winkler, M.; Martinkova, L.; Knall, A. C.; Krahulec, S.; Klempier, N. Tetrahedron 2005, 61, 4249.
 (c) Preiml, M.; Honig, H.; Klempier, N. J. Mol. Catal. B: Enzym. 2004, 29, 115. (d) Preiml, M.; Hillmayer, K.; Klempier, N. Tetrahedron Lett. 2003, 44, 5057. (e) Sugai, T.; Yamazaki, T.; Yokoyama, M.; Ohta, H. Biosci., Biotechnol., Biochem. 1997, 61, 1419.

^{(2) (}a) Groger, H. Chem. Rev. 2003, 103, 2795. (b) Enders, D.; Shilvock, J. P. Chem. Soc. Rev. 2000, 29, 359.

^{(3) (}a) Miller, J. A.; Nguyen, S. T. *Mini-Rev. Org. Chem.* **2005**, *2*, 39. (b) Reissig, H. U.; Zimmer, R. *Chem. Rev.* **2003**, *103*, 1151.

^{(4) (}a) Mangelinckx, S.; De Kimpe, N. *Synlett* **2006**, 369. (b) Mangelinckx, S.; De Kimpe, N. *Synlett* **2005**, 1521. (c) Mangelinckx, S.; De Kimpe, N. *Tetrahedron Lett.* **2003**, *44*, 1771.

⁽⁵⁾ D'hooghe, M.; Mangelinckx, S.; Persyn, E.; Van, Brabandt, W.; De Kimpe, N. J. Org. Chem. 2006, 71, 4232.

^{(6) (}a) Duke, R. K.; Chebib, M.; Balcar, V. J.; Allan, R. D.; Mewett, K. N.; Johnston, G. A. R. J. Neurochem. 2000, 75, 2602. (b) Johnston, G. A. R. Curr. Med. Chem. 2002, 2, 903. (c) Crittenden, D. L.; Park, A.; Qiu, Jian, S.; Richard, B.; Duke, R. K.; Johnston, G. A. R.; Jordan, M. J. T.; Chebib, M. Bioorg. Med. Chem. 2006, 14, 447. (d) Carland, J. E.; Moorhouse, A. J.; Barry, P. H.; Johnston, G. A. R.; Chebib, M. J. Biol. Chem. 2004, 279, 54153. (e) Carland, J. E.; Moore, A. M.; Hanrahan, J. R.; Mewett, K. N.; Duke, R. K.; Johnston, G. A. R.; Chebib, M. J. Biol. R.; Mewett, K. N.; Duke, R. K.; Johnston, G. A. R.; Chebib, M. J. Biol. Chem. 2004, 279, 54153. (e) Carland, J. E.; Moore, A. M.; Hanrahan, J. R.; Mewett, K. N.; Duke, R. K.; Johnston, G. A. R.; Chebib, M. Neuropharmacology 2004, 46, 770.

⁽⁷⁾ Foley, L. H.; Wang, P.; Dunten, P.; Ramsey, G.; Gubler, M.; Wertheimer, S. J. *Bioorg. Med. Chem. Lett.* **2003**, *13*, 3871.

⁽⁸⁾ Mallams, A. K.; Dasmahaptra, B.; Neustadt, B. R.; Demma, M.; Vaccaro, H. A. U.S. Pat. Appl. US 2007015774 A1, Jan 18, 2007; *Chem. Abstr.* **2007**, *146*, 163135.

⁽⁹⁾ Donghi, M.; Ferrara, M.; Koch, U.; Narjes, F.; Ontoria Ontoria, J. M.; Summa, V. PCT Int. Appl. WO 2007028789 A1, Mar 15, 2007; *Chem. Abstr.* **2007**, *146*, 316932.

in cyclopropane γ -peptides.¹⁰ Despite their biological relevance, very few synthetic approaches toward 2-(aminomethyl)cyclopropanecarboxylic acid derivatives are available in the literature. These methods comprise the cyclopropanation of allylic amines by means of diazoacetates,¹¹ the reaction of allylsulfonamides with phenyl(alkynyl)iodonium salts,¹² and the ring opening of 3-azabicyclo[3.1.0]hexan-2-one derivatives.¹³



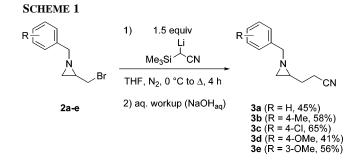
In the present report, the synthesis of 2-(2-cyanoethyl)aziridines is disclosed as a new and versatile class of building blocks in organic chemistry. Whereas the chemistry of 2-(cyanomethyl)aziridines has been studied previously for the preparation of different types of aminonitriles,^{5,14} the synthesis and reactivity of their higher homologues 2-(2-cyanoethyl)aziridines comprises an unexplored field of research, as only one similar compound has been reported so far, i.e., 1-butyl-2-(2-cyanoethyl)aziridine-2-carbonitrile obtained via cycloaddition of butyl azide with the appropriate olefin.¹⁵ In this paper, 1-arylmethyl-2-(2-cyanoethyl)aziridines were subsequently transformed into novel 2-[N,N-bis(arylmethyl)aminomethyl]cyclopropanecarbonitriles in an efficient and straightforward approach by means of an intramolecular 1,3-cyclization protocol of intermediate 5-amino-4-bromopentanenitriles, obtained through ring opening with benzyl bromide. This is the first report of the use of γ -halo- δ -aminopentanenitriles as substrates for a 3-*exo-tet* cyclization toward 2-(aminomethyl)cyclopropanecarbonitriles.

Results and Discussion

1-Arylmethyl-2-(bromomethyl)aziridines **2**, prepared from the corresponding benzaldehydes in a three-step procedure,¹⁶ are suitable synthetic equivalents for the aziridinylmethyl cation, providing an easy access to 2-substituted 1-(arylmethyl)aziridines upon treatment with carbon-centered as well as heteroatom-centered nucleophiles.¹⁷ Further elaboration of this approach resulted in the synthesis of 1-arylmethyl-2-(2-cyanoethyl)aziridines **3** as a novel class of compounds upon treatment of 2-(bromomethyl)aziridines **2** with 1.5 equiv of α -lithiated trimethylsilylacetonitrile (prepared by treatment of trimethyl-

2007, 72, 4733.

(15) Broeckx, W.; Overbergh, N.; Samyn, C.; Smets, G.; L'Abbe, G. Tetrahedron 1971, 27, 3527.



silylacetonitrile with an equimolar amount of *n*-BuLi in THF at 0 °C) in THF after reflux for 4 h (Scheme 1). Only by utilizing 1.5 equiv of α -lithiated trimethylsilylacetonitrile and heating under reflux for 4 h could the reaction be driven to completion. If fewer equivalents were used or shorter reaction times were applied, the reaction mixture contained a certain amount of starting material. The trimethylsilyl group was cleaved off from the initially formed silylated aziridine intermediate during workup by means of an aqueous NaOH solution (1 N). 1-Arylmethyl-2-(2-cyanoethyl)aziridines **3** were obtained in high purity by means of column chromatography on silica gel. The reactive nature of the constrained aziridine ring in 2-(2-cyanoethyl)aziridines **3** enables the preparation of a variety of new aminonitrile derivatives, complementary to the synthetic usefulness of 1-arylmethyl-2-(cyanomethyl)aziridines.

It should be noted that the use of methyl or ethyl trimethylsilyl acetate instead of trimethylsilylacetonitrile, applying either exactly the same reaction conditions or slightly modified conditions according to a literature procedure,¹⁸ did not result in any reaction, and the starting material was recovered completely each time. In the literature, only very few isolated examples of 2-(2-alkoxycarbonylethyl)aziridines have been reported, and no general approach is available toward these azaheterocycles.¹⁹

Treatment of 2-(2-cyanoethyl)aziridines **3** with 1 equiv of benzyl bromide in acetonitrile afforded novel 5-amino-4bromopentanenenitriles **5** in excellent yields after reflux for 5 h (Scheme 2). These δ -aminonitriles were purified by means of column chromatography on silica gel (hexane/EtOAc 9/1) in order to obtain analytically pure samples. In this transformation, benzyl bromide is responsible for both the activation of the aziridine ring toward an aziridinium intermediate **4** and the delivery of the nucleophilic bromide anion which induces ring opening of the aziridinium ion.

In accordance with the previously observed reactivity of 2-(bromomethyl)-, 2-(aryloxymethyl)-, 2-(alkanoyloxymethyl)-, and 2-(cyanomethyl)aziridines toward arylmethyl bromides in acetonitrile,^{14b,20} a regiospecific ring opening of the intermediate 2-(2-cyanoethyl)aziridinium salts **4** by bromide occurred at the

⁽¹⁰⁾ Qureshi, M.; Khurram, N.; Smith, M. D. Chem. Commun. 2006, 5006.

^{(11) (}a) Bergbreiter, D. E.; Morvant, M.; Chen, B. Tetrahedron Lett.
1991, 32, 2731. (b) Paulini, K.; Reisig H. Liebigs Ann. Chem. 1991, 455.
(12) Lee, H.-Y.; Lee, Y.-H. Synlett 2001, 1656.

⁽¹³⁾ Galeazzi, R.; Mobbili, G.; Orena, M. Tetrahedron: Asymmetry 1997, 8, 133.

^{(14) (}a) D'hooghe, M.; Vervisch, K.; Van Nieuwenhove, A.; De Kimpe, N. *Tetrahedron Lett.* **2007**, *48*, 1771. (b) D'hooghe, M.; Van Speybroeck, V.; Van, Nieuwenhove, A.; Waroquier, M.; De Kimpe, N. J. Org. Chem.

^{(16) (}a) De Kimpe, N.; Jolie, R.; De Smaele, D. J. Chem. Soc., Chem.
Commun. 1994, 1221. (b) De Kimpe, N.; De Smaele, D.; Szakonyi, Z. J.
Org. Chem. 1997, 62, 2448. (c) D'hooghe, M.; Waterinckx, A.; De Kimpe, N. J. Org. Chem. 2005, 70, 227. (d) D'hooghe, M.; Rottiers, M.; Jolie, R.; De Kimpe, N. Synlett 2005, 931.

^{(17) (}a) D'hooghe, M.; Rottiers, M.; Kerkaert, I.; De Kimpe, N. *Tetrahedron* **2005**, *61*, 8746. (b) D'hooghe, M.; Rottiers, M.; Jolie, R.; De Kimpe, N. *Synlett* **2005**, 931. (c) D'hooghe, M.; Kerkaert, I.; Rottiers, M.; De Kimpe, N. *Tetrahedron* **2004**, *60*, 3637. (d) D'hooghe, M.; Waterinckx, A.; Vanlangendonck, T.; De Kimpe, N. *Tetrahedron* **2006**, *62*, 2295. (e) D'hooghe, M.; De Kimpe, N. *ARKIVOC* **2008**, *(ix)*, *6.*

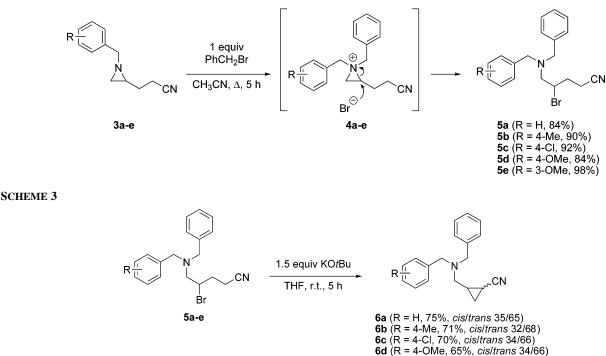
⁽¹⁸⁾ Gilday, J. P.; Gallucci, J. C.; Paquette, L. A. J. Org. Chem. 1989, 54, 1399.

⁽¹⁹⁾ Only very few isolated examples of 2-(2-alkoxycarbonylethyl)aziridines have been reported: (a) Vesely, J.; Ibrahem, I.; Zhao, G.-L.; Rios, R.; Cordova, A. Angew. Chem., Int. Ed. **2007**, 46, 778. (b) Fioravanti, S.; Pellacani, L.; Stabile, S.; Tardella, P. A.; Ballini, R. Tetrahedron **1998**, 54, 6169. (c) Ho, M.; Chung, J. K. K.; Tang, N. Tetrahedron Lett. **1993**, 34, 6513. (d) Gustavson, L. M.; Rao, T. N.; Jones, D. S.; Fritzberg, A. R.; Srinivasan, A. Tetrahedron Lett. **1991**, 32, 5485.

^{(20) (}a) D'hooghe, M.; Van Brabandt, W.; De Kimpe, N. J. Org. Chem.
2004, 69, 2703. (b) D'hooghe, M.; Waterinckx, A.; Vanlangendonck, T.; De Kimpe, N. Tetrahedron 2006, 62, 2295. (c) D'hooghe, M.; Van Speybroeck, V.; Waroquier, M.; De Kimpe, N. Chem. Commun. 2006, 1554. (d) D'hooghe, M.; De Kimpe, N. Synlett 2006, 2089.

JOC Article





more hindered aziridine carbon atom, affording β -bromoamines **5** as the sole reaction products in high purity. Detailed spectral analysis confirmed the structural identity of these novel *N*-(2-bromo-4-cyanobutyl)amines **5**, excluding the formation of the corresponding regioisomers.

It should be remarked that this novel approach offers a direct method toward a functionalized γ -bromonitrile moiety, which is a new structural entity suitable for further elaboration and difficult to prepare via other routes. Only one alternative approach toward 5-amino-4-bromopentanenitriles can be found in the literature, involving an exchange reaction of xanthates with bromine in refluxing chlorobenzene.²¹

The presence of this γ -bromonitrile moiety in aminonitriles 5 allows a formal 1,3-intramolecular ring closure toward cyclopropanecarbonitriles upon α -deprotonation with respect to the nitrile moiety and subsequent bromide expulsion through nucleophilic displacement. In accordance with the previously reported 3-exo-tet ring closure of N-[2-chloro-1-(cyanomethyl)ethyl]benzimidates toward N-(2-cyanocyclopropyl)benzimidates,⁵ the premised cyclization of γ -bromonitriles **5** proceeded very nicely utilizing 1.5 equiv of KOtBu in THF at room temperature for 5 h, affording novel 2-(aminomethyl)cyclopropanecarbonitriles 6 in good yields (Scheme 3). Almost no information regarding this type of compounds can be found in the literature,²² and no general synthetic approach toward 2-(aminomethyl)cyclopropanecarbonitriles is available to date. These constrained carbocycles can be considered as precursors of the corresponding biologically relevant 2-(aminomethyl)cyclopropanecarboxylic acids with diverse applications in medicinal and peptide chemistry.

Based on literature data, it is generally observed that the substituted cyclopropane carbon atoms in trans-disubstituted cyclopropane derivatives are characterized by a downfield shift in ¹³C NMR as compared to their cis isomers.^{5,23} Also in 2-(aminomethyl)cyclopropanecarbonitriles 6, the C2-carbon atoms (CHCH₂N) of the major isomers resonated at considerably higher δ values (19.1–19.2 ppm, CDCl₃) as compared to the minor isomers (17.0 ppm, CDCl₃), confirming the formation of trans-2-(aminomethyl)cyclopropanecarbonitriles 6 as the major constituents. The trans isomers of compounds 6a - e were obtained in pure form after separation by means of column chromatography on silica gel (hexane/EtOAc 14/1), allowing full spectroscopic characterization. The diastereomeric ratio of cyclopropanes 6 (cis/trans 32-35/65-68) was determined by means of ¹H NMR. This ratio appeared to be independent of the reaction temperature, as the same values were obtained upon treatment with 1.5 equiv of KOtBu in THF after reflux for 30 min or after stirring at 0 °C for 6 h.

6e (R = 3-OMe, 64%, cis/trans 33/67)

In conclusion, 1-arylmethyl-2-(bromomethyl)aziridines have been converted into 2-(2-cyanoethyl)aziridines upon treatment with α -lithiated trimethylsilylacetonitrile in THF in an efficient and straightforward approach. The latter novel aziridines can serve as substrates for the preparation of a variety of aminonitrile derivatives via elaboration of the aziridine moiety. 1-Arylmethyl-2-(2-cyanoethyl)aziridines were further transformed into 5-amino-4-bromopentanenitriles through a regiospecific ring opening of intermediate aziridinium salts upon treatment with benzyl bromide in acetonitrile. The latter γ -bromonitriles proved to be excellent substrates for the 1,3-intramolecular ring closure toward novel 2-(aminomethyl)cyclopropanecarbonitriles by means of KOtBu in THF. The net conversion of this methodology concerns a novel ring transformation of an aziridine into an (aminomethyl)cyclopropane derivative.

⁽²¹⁾ Barbier, F.; Pautrat, F.; Quiclet-Sire, B.; Zard, S. Z. Synlett 2002, 811.

^{(22) (}a) Grundstrom, K.; Dahlbom, R. Acta Pharm. Suec. 1972, 9, 491.
(b) Frydman, B.; Blokhin, A. V.; Brummel, S.; Wilding, G.; Maxuitenko, Y.; Sarkar, A.; Bhattacharya, S.; Church, D.; Reddy, V. K.; Kink, J. A.; Marton, L. J.; Valasinas, A.; Basu, H. S. J. Med. Chem. 2003, 46, 4586.

^{(23) (}a) Csuk, R.; Kern, A. *Tetrahedron* **1999**, *55*, 8409. (b) Theberge, C. R.; Verbicky, A. C.; Zercher, C. K. J. Org. Chem. **1996**, *61*, 8792.

Experimental Part

Synthesis of 1-Arylmethyl-2-(2-cyanoethyl)aziridines 3. General procedure: To an ice-cooled solution of trimethylsilylacetonitrile (7.5 mmol, 1.5 equiv) in dry THF (10 mL) was added n-BuLi (1.5 equiv, 2.5 M) via a syringe under nitrogen atmosphere, and the resulting solution was stirred for 1 h at 0 °C. Subsequently, a solution of 1-arylmethyl-2-(bromomethyl)aziridine 2 (5 mmol) in THF (10 mL) was added via a syringe at 0 °C, followed by heating under reflux for 4 h. The reaction mixture was poured into a 1 N NaOH_{aq} solution (25 mL) and extracted with Et₂O (3 \times 20 mL). Drying (MgSO₄), filtration of the drying agent, and evaporation of the solvent afforded 1-arylmethyl-2-(2-cyanoethyl)aziridine 3, which was purified by means of column chromatography on silica gel. 1-Benzyl-2-(2-cyanoethyl)aziridine 3a. Yellow oil. Yield: 45%. $R_f = 0.43$ (hexane/EtOAc/Et₃N 40/10/1). ¹H NMR (300 MHz, $CDCl_3$): δ 1.45–1.52 (1H, m); 1.56 (1H, d, J = 6.1 Hz); 1.60– 1.67 (1H, m); 1.75 (1H, d, J = 3.3 Hz); 1.83–1.94 (1H, m); 2.17 $(1H, d \times d \times d, J = 16.9, 7.9, 7.0 Hz); 2.26 (1H, d \times d \times d, J =$ 16.9, 7.2, 6.0 Hz); 3.22 and 3.64 (2H, $2 \times d$, J = 12.8 Hz); 7.26-7.39 (5H, m). ¹³C NMR (75 MHz, CDCl₃): δ 15.3 (CH₂), 29.0 (CH₂), 34.4 (CH₂), 37.5 (CH), 64.9 (CH₂), 119.5 (C), 127.5 (CH), 128. 5 (CH), 128.7 (CH), 138.9 (C). IR (NaCl, cm⁻¹): $\nu_{CN} = 2246$. MS (70 eV) m/z: 187 (M⁺ + 1, 100). Anal. Calcd for C₁₂H₁₄N₂: C, 77.38; H, 7.58; N, 15.04. Found: C, 77.54; H, 7.77; N, 14.93.

Synthesis of 5-[Bis(arylmethyl)amino]-4-bromopentanenitriles 5. General procedure: To a stirred solution of 1-arylmethyl-2-(2cyanoethyl)aziridine 3 (5 mmol) in acetonitrile (10 mL) was added benzyl bromide (5 mmol, 1.0 equiv), and the resulting mixture was heated under reflux for 5 h. Evaporation of the solvent afforded 5-[bis(arylmethyl)amino]-4-bromopentanenitrile 5, which was purified by means of column chromatography on silica gel (hexane/ EtOAc 9/1) in order to obtain an analytically pure sample. 5-[Bis(benzyl)amino]-4-bromopentanenitrile 5a. Yellow oil. Yield: 84%. $R_{\rm f} = 0.24$ (hexane/EtOAc 9/1). ¹H NMR (300 MHz, CDCl₃): δ 1.75–1.88 (1H, m); 2.12–2.40 (3H, 2 × m); 2.78 and 2.88 (2H, 2 × d × d, J = 13.4, 9.5, 5.4 Hz); 3.49 and 3.72 (4H, $2 \times d$, J = 13.2 Hz); 3.91-3.99 (1H, m); 7.22-7.38 (10H, m). ¹³C NMR (75 MHz, CDCl₃): δ 15.1 (CH₂), 31.5 (CH₂), 50.9 (CH), 59.6 (CH₂), 60.1 (CH₂), 119.1 (C), 127.6 (CH), 128.6 (CH), 129.2 (CH), 138.7 (C). IR (NaCl, cm⁻¹): $\nu_{CN} = 2250$. MS (70 eV) m/z: 357/9 (M⁺ + 1, 24), 277 (100). Anal. Calcd for $C_{19}H_{21}BrN_2$: C, 63.87; H, 5.92; N, 7.84. Found: C, 64.02; H, 6.05; N, 7.71.

Synthesis of 2-[N,N-Bis(arylmethyl)aminomethyl]cyclopropanecarbonitriles 6. General procedure: To a solution of 5-[bis-(arylmethyl)amino]-4-bromopentanenitrile 5a (2 mmol) in dry tetrahydrofuran (15 mL) was added potassium tert-butoxide (3 mmol, 1.5 equiv) at room temperature. After being stirred for 5 h at room temperature, the reaction mixture was poured into water (30 mL) and extracted with diethyl ether (3 \times 25 mL). The combined organic extracts were washed with water (2 \times 50 mL) and brine $(1 \times 50 \text{ mL})$. Drying (MgSO₄), filtration of the drying agent, and removal of the solvent afforded 2-[N,N-bis(arylmethyl)aminomethyl]cyclopropanecarbonitrile 6 as a mixture of cis/trans isomers. The trans isomer was isolated in pure form by means of column chromatography on silica gel (hexane/EtOAc 14/1). trans-2-[N,N-Bis(benzyl)aminomethyl]cyclopropanecarbonitrile trans-**6a.** Colorless oil. $R_f = 0.15$ (hexane/EtOAc 14/1). ¹H NMR (300 MHz, CDCl₃): δ 0.65-0.72 (1H, m); 0.82-0.95 (1H, m); 1.10-1.28 (1H, m); 1.53-1.63 (1H, m); 2.33 and 2.38 (2H, $2 \times d \times d$, J = 13.8, 6.6, 6.1 Hz); 3.60 and 3.64 (2 × 2H, 2 × d, J = 13.6Hz); 7.19–7.37 (10H, m). ¹³C NMR (75 MHz, CDCl₃): δ 2.4 (CH), 13.2 (CH₂), 19.2 (CH), 55.6 (CH₂), 58.6 (CH₂), 121.7 (C), 127.3 (CH), 128.6 (CH), 129.8 (CH), 139.3 (C). IR (NaCl, cm⁻¹): ν_{CN} = 2238. MS (70 eV) m/z: 277 (M⁺ + 1, 100). Anal. Calcd for C₁₉H₂₀N₂: C, 82.57; H, 7.29; N, 10.14. Found: C, 82.73; H, 7.44; N, 10.07. cis-2-[N,N-Bis(benzyl)aminomethyl]cyclopropanecarbonitrile *cis*-6a. Colorless oil. $R_f = 0.05$ (hexane/EtOAc 14/1). ¹H NMR (300 MHz, CDCl₃): δ 0.73-0.79 (1H, m); 0.83-0.90 (1H, m); 1.10-1.17 (1H, m); 1.35-1.42 (1H, m); 2.61 and 2.77 (2H, 2 × d × d, J_{gem} = 13.4, 5.6, 5.0 Hz); 3.65 and 3.72 (2 × 2H, 2 × d, J = 13.8 Hz); 7.21–7.42 (10H, m). ¹³C NMR (75 MHz, CDCl₃): δ 2.4 (CH), 13.3 (CH₂), 17.0 (CH), 54.9 (CH₂), 58.7 (CH₂), 120.3 (C), 127.1 (CH), 128.4 (CH), 128.8 (CH), 139.5 (C). IR (NaCl, cm⁻¹): $\nu_{\rm CN} = 2238$. MS (70 eV) m/z: 277 (M⁺ + 1, 100).

Acknowledgment. We are indebted to the "Fund for Scientific Research - Flanders (Belgium)" (F.W.O.-Vlaanderen) and to Ghent University (GOA) for financial support.

Supporting Information Available: Spectroscopic data of compounds **3b–e**, **5b–e**, and **6b–e**. This material is available free of charge via the Internet at http://pubs.acs.org.

JO701302A